CARBON CONTENT (ORGANIC AND TOTAL)

BY

PATRICIA J. BLAKESLEE

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EXPLANATION

Introduction

Many chemical reactions

Many chemical reactions occurring in the Chesapeake Bay estuary depend upon the availability of organic carbon. In addition, the concentration of this element serves as a pollution level indicator and aids in the location of sites with potentially high concentrations of heavy metals and/or other polluting substances.

Carbon is the primary food source for organisms in the Chesapeake

Bay. Carbon, whether in the form of organic matter or living organisms, is extracted from the water column and sediment by organisms and, in the presence of oxygen, is 'burned' to provide energy. Unless oxygen is constantly replenished, it is eventually depleted by this process. The environment then becomes anoxic, and the consumption of organic carbon decreases dramatically. Many of the fine-grained sediments of the Bay bottom become anoxic within a few centimeters of the sediment surface. Within these anoxic sediments, reactive metals, such as iron and manganese, combine with reduced sulfur species to form metal sulfides. These are stable in the sediment as long as the environment remains anoxic. However, if these sediments are disturbed and introduced into an oxidizing environment, the following could occur: 1) the creation of an oxygen demand proportional to the concentrations of organic carbon and reduced compounds in the sediment; 2) the formation of oxidation products analogous to those found in acid mine drainage, as a result of oxidation of the iron sulfide phases; or 3) the release of nutrients and trace metals into the environment. Knowing the organic carbon content of the sediments can play a key role in identifying areas of anoxic sediments and estimating potentially deleterious effects should these sediments be disturbed.

Carbon is also present in the sediments as mineralic skeletal parts, such as shell, teeth, and bones. These materials are viewed as inorganic because they cannot be utilized as an energy or food source by organisms, nor do they affect the potential oxygen demand of anoxic sediments. Although they do contribute to the total amount of carbon present in the sediment, their contribution is negligible. The amount of total carbon in any particular sediment sample is largely controlled by the amount of organic carbon in that sample. For this reason, the remainder of the discussion concerns the organic carbon distribution in the sediments, and the contour lines of equal carbon content shown on the map refer to organic carbon.

Carbon analysis was done on approximately one out of every four samples collected from the deeper water of the Bay. Sediments consisting predominantly of sand, which generally occur in shallower water, contain amounts of organic carbon close to or below the detection level of the analytical equipment. Therefore, these samples were not analyzed for their carbon content. Both total and organic carbon content were determined for 27 samples using a LECO Gasometric Analyzer (Model #572-100) in conjunction with a LECO Induction Furnace (Model #521-000).

Distribution

Areas of high organic carbon content tend to correspond to areas of deep water and fine-grained sediment. Nearshore and beach areas are high energy, wave-dominated zones in which constant reworking of the sediments results in the removal of the fine-grained materials, including the organics. In addition, high energy conditions stir up the bottom, aerating the sediments and preventing anoxic conditions from developing. In contrast, deep areas tend to be low energy environments in which fine-grained materials, including organics, accumulate. Anoxic conditions develop quickly in these materials because their fine-grained nature inhibits the passage of oxygen into the sediment from the overlying water. Because anaerobic decomposition of organic materials is a much slower process than aerobic decomposition, organic carbon material is preserved in these sediments.

Dry weight percent organic carbon in this section of the Bay averages 5.21%, ranging from 0.61% to 10.47% (Table 1). The highest concentrations of organic carbon known to occur in the Maryland portion of Chesapeake Bay are found here, within a lobe of relatively coarse sediments deposited immediately downbay of the mouth of the Susquehanna River. The exceptionally high carbon content of these sediments is related to the terrigenous origin of the carbon, which includes particulate coal and plant debris. Coal from mining operations in northeastern Pennsylvania is transported to the head of the Bay by the Susquehanna (Ryan, 1953; Folger, 1972; Williams and Reed, 1972; Palmer, 1975). Most of the sediment load of the river is deposited within this region. The proportion of sedimentary carbon derived from terrigenous sources steadily decreases from the head of the Bay, with primary productivity contributing progressively more of the carbon incorporated into the sediments. As the nature of the carbon changes, the common association of organic carbon with fine-grained sediments is better established.

RANGE % C	MEAN % C	NUMBER
0.61-0.98 4.15-10.13 (0.61-10.13)	0.80 7.14 - (3.97)	2 2 (4)
3.34-6.46 (3.34-6.46)	4.58 (4.58)	11 (11)
	=	-
1.71-10.47	6.19	12
0.61-10.47	5.21	27
	0.61-0.98 4.15-10.13 (0.61-10.13) - 3.34-6.46 (3.34-6.46)	0.61-0.98 4.15-10.13 (0.61-10.13) (3.97)

Folger, D.W., 1972, Characteristics of Estuarine Sediments of the United States: U.S. Geol. Survey Prof. Paper 742, 94 p.

Palmer, H.D., 1975, Bottom sediments, in Munson, T.O., ed., Upper Bay Survey: Annapolis, Md., Westinghouse Electric Corp., vol. II, p. 4-1 to 4-33.

Ryan, J.D., 1953, The sediments of Chesapeake Bay: State of Maryland, Dept. of Geology, Mines and Water Resources, Bull. 12, 120 p.

Williams, K.F., and Reed, L.A., 1972, Appraisal of stream sedimentation in the Susquehanna River basin: U.S. Geol. Survey Water Supply Paper 1532-F, 24 p.

LEGEND

1.5 - ORGANIC CARBON % DRY WEIGHT
(1.7) - TOTAL CARBON % DRY WEIGHT
CONTOUR INTERVAL 1% ORGANIC CARBON

